MARK SCHEME for the May/June 2010 question paper

for the guidance of teachers

5129 COMBINED SCIENCE

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



UNIVERSITY of CAMBRIDGE International Examinations

	Page 2		Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL – May/June 2010	5129	02
1	(a)	correct s in paralle	ymbol I with 6Ω resistor		[2]
	(b)	(i) ∨ = = 1.2	R (or equivalent) or 0.2 × 6		
		V	(unit independent)		[3]
		(ii) 9			[1]
2	(a)	metals	um and aluminium / Mg and A <i>l</i> are independent)		[2]
	(b)		rple / violet nge / yellow		[2]
	(c)		mber of / 2 electrons in outermost shell lence electrons)		[1]
3	(a)	(i) 0 an	d 1		[1]
		(ii) 2 an	d 3		[1]
		(iii) 3 an	d 4		[1]
	(b)	iris / circi	ular and/or radial muscles		[1]
	(c)		ich remains above the drawn line throughout along initial line but must not increase)		[1]
	(d)	retina			[1]
4	(a)	27			[1]
	(b)	electron	/ e		[1]
	(c)	32			[1]
	(d)	= 11 400	tion of 2 half lives with incorrect answer = 1 mark		[2]

	Page 3		Mark Scheme GCE O LEVE			Syllabus 5129	Paper 02
5		liver amino ac kidneys liver					[4]
6	(a)	neutralis	ation				[1]
	(b)	17 80					[2]
	(c)	(17 × 2) = 0.425 k (ecf from	kg / 425 g				[2]
7	(a)		ack is a better emitter / ra / white is a poorer emitte				[1]
	(b)	conduction	on				[1]
	(c)		on air expands air is less dense	}	any one		[1]
8	(a)	filter add chlo sedimen		}	any two		[2]
	(b)	kill bacte	solids / insoluble particle eria / micro-organisms ettle to the bottom	s / dirt			[2]
9	(a)		sade / mesphyll er) epidermis				[2]
	(b)	<u>spongy r</u>	<u>mesophyll</u>				[1]
	(c)	by diffusi through s					[2]
	(d)		over short distance / ea netrates to all parts / more				[2]

	Page 4		Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL – May/June 2010	5129	02
10	(a)	no yes n 4 = 2 ma	o yes rks 3 = 1 mark		[2]
	(b)	(i) plas	tic		[1]
		(ii) iron	/ Fe		[1]
11	(a)	haematit	e / magnetite		[1]
	(b)	(i) 3 (allo	2 3 w correct multiples)		[1]
		(ii) remo	oval of oxygen / gain of electrons / lower oxidation state	e	[1]
		Carb	e / carbon burns to form carbon dioxide on dioxide reacts with carbon on reacts with oxygen = 1 mark		[2]
	(c)	too react	ive / more reactive than iron / carbon		[1]
12	(a)	(i) 60			
		(ii) 8 (ig	nore any sign)		[2]
	(b)		n / disturbance / displacement / motion of particles cular to the direction / motion of the wave		[2]
13	(a)	crush / te	wn of large pieces to small pieces / ear / grind food		
		mix food make sw	surface area with saliva / enzymes vallowing easier eferences to chewing / digestion)		[2]
	(b)		ctly identified) cayed teeth / less decay		[1]
	(c)	brush tee visit dent use more	sugar ter dental care / more clinics / dentists eth more tist more regularly e mouthwash		
		eat more better ed	fruit and vegetables ucation		[2]

accept converse in terms of town A

	Page 5		Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL – May/June 2010	5129	02
14	(a)	infra red	/ ir		[1]
	(b)	radio			[1]
	(c)	300 000	000 / 3 × 10 ⁸		[1]
15	(a)	carbon d	ioxide		
		plan	ts animals		[3]
	(b)		osynthesis		
		Y = respi Z = deca	ration y / decomposition / (bacterial) respiration		[3]
16	(a)	D			[1]
	(b)	В			[1]
	()				
	(c)	С			[1]
17	(a)	time = di 4.0	stance/speed or 0.8/0.2		[2]
	(b)	work = fc 3.2	prce × distance or 4 × 0.8		
		J / joules	(unit independent)		[3]
18	(a)	carbon d	ioxide team (any order)		[2]
					[~]
	(b)	carbon n poisonou	nonoxide is / toxic / correct description of mode of action		[2]
	(c)		neral formula		[1]
		gradatior	hemical properties in physical properties differ by CH ₂ } any one		[1]

	Page 6				Paper
			GCE O LEVEL – May/June 2010	5129	02
19	(a)		/ gravitational movement		[2]
	(b)	0.5			[1]
20	(a)	(i) any	date from Feb 28 to Mar 5		[1]
		(ii) any	date from Mar 10 to Mar 17		[1]
	(b)		lays / 4 weeks		[1]
		stre diet preg exe	/ menopause ss / emotional state / anxiety / malnutrition / starvation gnancy / breast feeding rcise etic factors / inheritance		[2]
21	(a)	fermenta	ation / anaerobic respiration		[1]
	(b)	(i) prov	vides enzymes		[1]
		(ii) prev	vents oxidation of ethanol / formation of ethanoic acid		[1]
	(c)	(fraction	al) distillation		[1]
	(d)	correct s	structure of ethanol		[1]